

[=>Back To Characterization Lab](#)

[=>Back To Polymer Processing](#)

Download this page: [=>Lab 7.pdf](#)

Download Excel Data for PDMS centistoke versus g/mole: [=>Lab7.xls.hqx](#)

## Viscosity Analysis of Polymers

### Objective:

The objective of this lab is to become familiar with the use of a viscometer in the determination of the extent of reaction in a polymerization reaction using a Brookfield Couette Viscometer (cup in bob viscometer). The experiment involves the condensation polymerization of hydroxyl terminated polydimethylsiloxane (PDMS) using a tin catalyst at 80 degrees C.

### Instruments to be used:

Brookfield Viscometer (Departmental)

Viscometer is Located in Old SEM Lab Across From Doug's Office In Rhodes 5'th Floor

### Materials:

Hydroxy- terminated PDMS, tin catalyst

### Procedure:

- 1) Add 5% tin catalyst to PDMS and mix.
- 2) Pour the mixture into the viscometer cell and measure the viscosity as a function of time until termination of the reaction.
- 3) *Repeat the experiment for several different concentrations catalyst and for several molecular weight partial condensates.*
- 4) With the final polymer you should measure the viscosity at several different rotational viscosities (different shear rates).

### Analysis:

- 1) Plot viscosity versus time for the reaction mixtures.
- 2) Use interpolations from the table of molecular weight versus melt viscosity at 80 degrees C to convert these plots to molecular weight versus time. (Doug or Sathish have this table).
- 3) Use the 3.4 power law dependence of zero shear rate viscosity versus molecular weight to determine the zero shear rate viscosity parameters for PDMS at high molecular weight.
- 4) Use the 1 power-law dependence in molecular weight at low molecular weight to determine the zero shear rate power law viscosity parameters.
- 5) *Look up literature values for these power-law fluid parameters in PDMS (Polymer Handbook or Processing books).*
- 6) Determine the time at which the entanglement molecular weight is reached for each reaction mixture.
- 7) From item #4 above construct a plot of viscosity versus rate of strain for the final polymer melt.

**Questions:**

- 1) Is the shear rate used in this experiment below the Newtonian plateau limit?
- 3) How is the melt viscosity determined in a Couette viscometer such as this Brookfield viscometer?
- 4) How does the melt viscosity of a polymer vary with shear rate?
- 5) *What are the power-law fluid parameters for PDMS (viscosity versus shear rate)?*