# Poisson's ratio and the fragility of glass-forming liquids

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The nature of the transformation by which a supercooled liquid 'freezes' to a glass—the glass transition—is a central issue in condensed matter physics<sup>1-3</sup> but also affects many other fields, including biology<sup>4</sup>. Substantial progress has been made in understanding this phenomenon over the past two decades, yet many key questions remain. In particular, the factors that control the temperature-dependent relaxation and viscous properties of the liquid phase as the glass transition is approached (that is, whether the glass-forming liquid is 'fragile' or 'strong'<sup>5-7</sup>) remain unclear. Here we show that the fragility of a glass-forming liquid is intimately linked to a very basic property of the corresponding glass phase: the relative strength of shear and bulk moduli, or Poisson's ratio.

Fragility of liquids is defined as the apparent activation energy of shear viscosity  $\eta$  or structural relaxation time  $\tau_{\alpha}$  at the glass transition temperature  $T_{g}$ , normalized to  $T_{g}$  (ref. 7):

$$m = \frac{\partial \log \eta(T)}{\partial (T_g/T)}\Big|_{T=T_g}$$
(1)

It characterizes the steepness of the slope of  $\log \eta$  dependence on  $T_g/T$  near  $T_g$  (Fig. 1). A stronger deviation from Arrhenius behaviour corresponds to a more fragile system. One of the puzzles that remains unsolved is the correlation of the fast terahertz dynamics and fragility<sup>8,9</sup>: one can predict fragility, that is, temperature variations of  $\tau_{\alpha}$  or  $\eta$  in a supercooled liquid at  $T \approx T_g$ , from analysis of vibrational spectra at terahertz frequencies deep in the glassy state ( $T \ll T_g$ )<sup>8,9</sup>. Interest in this puzzle was recently stimulated by ref. 9 (see also refs 10 and 11). Our present goal is to gain better insight into fragility on the microscopic scale and to suggest a consistent explanation for these correlations.

It is well understood that the glass transition is a failure of the material to support shear stress on the typical laboratory timescale of, say, a few minutes. So, shear modulus G is apparently an important parameter. To be more precise, the modulus is frequency-dependent and one needs to consider G at frequencies much higher than the inverse structural relaxation time, that is, the so-called instantaneous shear modulus  $G_{\infty}$ . In an isotropic solid like glass there is only one different independent elastic constant that makes it possible to get the dimensionless parameter proportional to  $G_{\infty}$ , namely, the instantaneous bulk modulus  $K_{\infty}$ . The ratio of instantaneous shear and bulk moduli can be estimated from the ratio of longitudinal  $(v_1^2 = M_{\infty}/\rho)$  and transverse  $(v_t^2 = G_{\infty}/\rho)$ sound velocities in the glassy state, for example,  $K_{\infty}/G_{\infty} \propto (v_1/v_1)$  $v_t$ )<sup>2</sup> - 4/3 (here  $\rho$  is the mass density and  $M_{\infty} = K_{\infty} + (4/3)G_{\infty}$  is the instantaneous longitudinal modulus). We note that the highfrequency sound velocity in glasses is connected with the adiabatic bulk modulus, although the difference between the isothermal and adiabatic bulk moduli in the glassy state is very small. Analysis of a large number of glasses, including covalent and hydrogen-bonded, van-der-Waals and ionic glasses, shows a correlation between  $v_1/v_1$ and m (Fig. 2): the more weakly the system resists the shear stress in comparison with the bulk one in the glassy state (higher  $v_1/v_1$ ), the more fragile its behaviour appears in the melt. We emphasize that the data presented in Fig. 2 includes all the systems we were able to find in the literature, excluding polymers. It is known that the fragility of some polymers strongly depends on molecular weight,  $M_w$  (ref. 12). For example, the fragility of polystyrene varies from 60 at low  $M_w$  up to 150 at high  $M_w$  (ref. 12), and only low- $M_w$  polystyrene is in agreement with the correlation of Fig. 2. This sharp increase in fragility with  $M_w$  might be polymer-specific; for this reason, all polymers were excluded from consideration, although low fragility polymers like polyisobutylene and polybutadiene agree well with the correlation of Fig. 2.

The inset in Fig. 2 shows the correlation of fragility m with  $K_{\infty}/G_{\infty}$  of the respective glass; it can be well described by the relationship:

$$m - 17 = 29(K_{\infty}/G_{\infty} - 1)$$
 or  $m = 29(K_{\infty}/G_{\infty} - 0.41)$  (2)

where 17 is the lowest value expected for m and  $K_{\infty}/G_{\infty}$  in glasses is not expected to be lower than 1 (for strong glasses like silica and BeF<sub>2</sub>,  $K_{\infty}/G_{\infty} \approx 1.1$ ). We note that correlation (equation (2)) means also that fragility of a liquid is fully determined just by the Poisson ratio of its glass.

How can one explain the correlation between the fragility of a liquid and the ratio of the instantaneous elastic moduli of a glass (Fig. 2)? The answer might be buried deep in the liquid state because the structure and properties of a glass are essentially the structure and properties of a 'frozen' liquid. Owing to the construction of the fragility plot, all viscosity or relaxation time curves intersect at two points: (1) at  $T_g$ ,  $\log \eta \equiv \log \eta_g = 13$  (where  $\eta$  is in poise), or  $\log \tau_{\alpha} \approx 3$  (where  $\tau$  is in seconds); and (2) at very high temperatures,  $T_g/T \rightarrow 0$ , where all liquids have  $\log \eta \equiv \log \eta_0 \approx -4$  (ref. 13) or  $\log \tau_{\alpha} \approx \log \tau_0 \approx -14$ . This means that if a liquid has a steeper slope of  $\log \eta$  near  $T_g$ , it inevitably has a smaller slope of  $\log \eta$  at high temperature. At high temperatures, relaxation in most of the liquids shows Arrhenius temperature dependence:  $\eta = \eta_0 \exp(E/T)$ . Thus the high-temperature slope of  $\log \eta$  in Fig. 1,  $E/T_g$ , can also be a measure of fragility. Experimental data show that  $E/T_g$  indeed correlates with fragility m (Fig. 3):

$$\frac{E}{T_{\rm g}} = \frac{(19.2)^2 \ln 10}{m}$$
(3)

We note that a similar relationship with a slightly different coefficient  $(17)^2 \ln 10$  follows from the Vogel–Fulcher–Tamman (VFT) equation,  $\eta = \eta_0 \exp B/(T - T_0)$ , if one assumes that the VFT equation is valid over the entire temperature range. The quantitative disagreement can be related to the well-known fact that a single VFT equation cannot accurately describe  $\eta$  at all temperatures<sup>14</sup>.



**Figure 1** An example of a fragility plot. Data is from refs 2, 27 and 28. DGG1 is a sodalime silica glass; see Supplementary Table 1. The solid lines show the expected hightemperature slopes  $E/T_{g}$ .

## letters to nature

This connection between *m* and  $T_g/E$  helps to explain the correlation between *m* and the ratio of instantaneous elastic moduli in the glassy state, equation (2). Various authors<sup>6,15,16</sup> suggest that the activation energy of viscosity is proportional to the instantaneous shear modulus  $G_{\infty}$ , that is,  $E \propto G_{\infty}V_c$ , where  $V_c$  is some volume that does not show significant temperature variations at high *T*. On the other hand, there are empirical correlations between elastic moduli of a glass and  $T_g$  (for example, refs 6, 17 and 18). However, it is not obvious which combination of shear and bulk moduli is important because these correlations were usually considered within a class of materials with similar chemical structure, and thus with a similar Poisson's ratio. We note that the free volume model for the glass transition<sup>18</sup> predicts  $T_g \propto K_{\infty}$ . Analysis of experimental data reveals correlation of  $T_g/E$  to the same parameter  $(K_{\infty}/G_{\infty} - 0.41)$  as in equation (2):

$$T_{\rm g}/E \approx 0.037 (K_{\infty}/G_{\infty} - 0.41) \propto m \tag{4}$$

Thus, the correlation found (equations (2) or (4) and Fig. 2) emphasizes a very simple rule: the better the glass can resist shear deformation rather than dilatation, the stronger (that is, more Arrhenius-like) the behaviour that is exhibited during its structural relaxation.

On a timescale longer than that of structural relaxation,  $t \gg \tau_{\alpha}$ , the shear modulus (for non-polymeric systems) relaxes to zero while the longitudinal modulus relaxes to some finite value  $M_0$ . This leads to a difference between instantaneous  $(\nu_{\infty})$  and zero-frequency  $(\nu_0)$ longitudinal sound velocities in glasses. The difference provides an estimate of the relaxation strength of the  $\alpha$ -process, that is, the socalled non-ergodicity parameter,  $f_0 = 1 - \nu_0^2/\nu_{\infty}^2$ . A sizeable part of the decrease of the longitudinal sound velocity at  $\omega \tau_{\alpha} \ll 1$  is due to relaxation of the shear modulus. Because M = K + (4/3)G, a crude approximation then gives:

$$f_0 \approx (4/3)v_t^2 / v_l^2 = G_\infty / [(3/4K_\infty + G_\infty]$$
 (5)

Thus, the ratio  $G_{\infty}/K_{\infty}$  also determines the amplitude of the nonergodicity parameter.

Equations (5) and (4) lead us directly to an explanation of the



**Figure 2** Correlation between fragility and the ratio of longitudinal and transversal sound velocities found in the glassy state. Inset, correlation of fragility with the ratio of the bulk and shear moduli in the glassy state. The straight line shows the relationship from equation (2). Materials (in descending order of fragility): CKN (Ca<sub>2</sub>(NO<sub>3</sub>)<sub>2</sub>-KNO<sub>3</sub>), m-TCP (m-tricresyl phosphate), OTP (o-therphenyl), Se, m-toluidine, salol, glycerol, B<sub>2</sub>O<sub>3</sub>, As<sub>2</sub>S<sub>3</sub>, 20(Na<sub>2</sub>O)80(SiO<sub>2</sub>), SiO<sub>2</sub>, GeO<sub>2</sub>, BeF<sub>2</sub>. In those cases where more than one value of fragility is known, we used the average value. Data from the literature for fragility,  $v_t$  and  $v_l$  and the respective references are given in Supplementary Table 1.

recent puzzling observation reported in ref. 9. The authors found that the parameter  $\alpha$ , defined as  $\alpha = R_{\text{LP}}^{-1}(T)T_g/T$ , measured using inelastic X-ray scattering in a few glasses, correlates with their fragility,  $\alpha \propto m$ . (Here  $R_{\text{LP}}(T) = I_c/2I_B$  is the Landau–Placzek ratio, and  $2I_{\text{Br}}$  and  $I_c$  are the integrated intensities of the combined Brillouin doublet and the central line of the dynamic structure factor  $S(q,\omega)$ , respectively.) It is known<sup>10,19</sup> that in the case of planewave phonons  $R_{\text{LP}}(T) = \kappa_T M_{\text{Br}} - 1$ , where  $\kappa_T$  is the isothermal compressibility, and  $M_{\text{Br}}$  is the adiabatic longitudinal modulus at the Brillouin line frequency. Using the relationships  $v_0^2 \approx 1/\rho\kappa_T$  and  $v_{\infty}^2 \approx M_{\infty}/\rho$ ,  $\alpha$  can be expressed via the non-ergodicity parameter at  $T_g$ :

$$\alpha = (1 - f_0)/f_0 \tag{6}$$

 $(1 - f_0)/f_0$  indeed correlates well with fragility<sup>10</sup>, as also does the non-ergodicity parameter  $f_0$ . Substituting equation (5) into equation (6) gives  $\alpha \propto K_{\infty}/G_{\infty}$ . This provides a clear microscopic interpretation of the correlation between  $\alpha$  and *m* reported in ref. 9:  $G_{\infty}/K_{\infty}$  controls both the relative amplitude of the structural relaxation in a glass and the fragility of a supercooled liquid.

 $f_0(T_g)$  also determines the amplitude of fluctuations frozen at  $T_g$ . That leads us to another unsolved problem-correlation between fragility and the amplitude of the boson peak<sup>20</sup>. All disordered materials have an excess density of vibrational states,  $g(\nu)$ , over the Debye density of states,  $g_{\text{Deb}}(\nu)$ , in the terahertz frequency range<sup>20–24</sup>, so that the ratio  $g(\nu)/g_{\text{Deb}}(\nu)$  exhibits the so-called boson peak with amplitude  $A_{bp}$ . According to refs 20–25,  $A_{bp}$  is related to the amplitude of frozen structure fluctuations. So, one would expect higher  $A_{\rm bp}$  in materials with higher  $f_0(T_{\rm g})$  and thus  $A_{\rm bp}$  should correlate with  $v_t/v_l$ , and, consequently, with fragility. Comparison of  $(v_t/v_l)^2$  and  $A_{bp}$  in a few glasses indeed shows very good correlation (Supplementary Fig. 1). Thus, the correlation of the boson peak amplitude to fragility observed in ref. 20 seems to be related to the same role of the non-ergodicity parameter. In other words, the capability of a structure to resist shear rather than bulk deformation also determines the boson peak amplitude through the amplitude of frozen fluctuations,  $f_0(T_g)$ .

Recently it was shown that higher anharmonicity of a glassformer corresponds to higher fragility<sup>8,11,26</sup>. To understand this, we note that anharmonicity always leads to an increase of the fast relaxation in glasses<sup>26</sup>, and thus, to a decrease of  $f_0$ . Because, as we showed above,  $f_0$  correlates with fragility, the same should be valid



**Figure 3** Correlation between fragility and the ratio of the high-temperature activation energy to the glass transition temperature. Materials are listed in ascending fragility order, where NBS717 is borosilicate glass, NBS711 is lead silica glass, BSC is borosilicate crown glass, TNB is 1,3,5-tri- $\alpha$ -naphthyl benzene and prop. is propylene. Data from the literature for fragility, the high-temperature slope of the fragility plot, *E*/*T*<sub>g</sub>, and the respective references are given in Supplementary Table 1.

for the anharmonicity. A similar explanation can be applied to the correlation between the intensity of the fast relaxation normalized to the boson peak and the fragility found in ref. 8. The amplitude of the fast relaxation is related to  $1 - f_0$ , so the ratio of the fast relaxation to the boson peak should be proportional to  $(1 - f_0)/f_0$ , and correlate linearly with  $\alpha$  and m.

Thus the ratio of instantaneous shear to bulk modulus in glasses, or, alternatively, the Poisson ratio, appears to be an important parameter that controls the fragility of liquids. This ratio also determines the non-ergodicity parameter  $f_0$ , that is, how limited is the motion of a molecular unit in a cage formed by its neighbours. The same non-ergodicity parameter controls behaviour of the fast dynamics and that explains the correlation between fragility and fast dynamics in glass-forming systems.

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## Low-voltage organic transistors with an amorphous molecular gate dielectric

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Organic thin film transistors (TFTs) are of interest for a variety of large-area electronic applications, such as displays<sup>1-3</sup>, sensors<sup>4,5</sup> and electronic barcodes<sup>6-8</sup>. One of the key problems with existing organic TFTs is their large operating voltage, which often exceeds 20 V. This is due to poor capacitive coupling through relatively thick gate dielectric layers: these dielectrics are usually either inorganic oxides or nitrides<sup>2-8</sup>, or insulating polymers<sup>9</sup>, and are often thicker than 100 nm to minimize gate leakage currents. Here we demonstrate a manufacturing process for TFTs with a 2.5-nm-thick molecular self-assembled monolayer (SAM) gate dielectric and a high-mobility organic semiconductor (pentacene). These TFTs operate with supply voltages of less than 2 V, yet have gate currents that are lower than those of advanced silicon field-effect transistors with SiO<sub>2</sub> dielectrics. These results should therefore increase the prospects of using organic TFTs in low-power applications (such as portable devices). Moreover, molecular SAMs may even be of interest for advanced silicon transistors where the continued reduction in dielectric thickness leads to ever greater gate leakage and power dissipation.

The use of SAM gate dielectrics for organic TFTs was pioneered by the Vuillaume group<sup>10</sup>. They investigated the mechanism of carrier tunnelling through SAMs and showed that current leakage through densely packed organic molecular monolayers with highly ordered aliphatic chains can be remarkably low, despite their thickness of only a few nanometres<sup>11</sup>. To prepare organic TFTs (and even silicon metal–insulator–semiconductor field-effect transistors, MISFETs) with SAM dielectrics, the Vuillaume group relied on carboxyl-terminated *n*-alkyltrichlorosilanes<sup>10,12</sup>. Using  $\alpha$ -sexithiophene as the organic semiconductor, they demonstrated organic TFTs operating at 2 V, with a subthreshold swing of 350 mV per decade, a field-effect mobility of  $3.6 \times 10^{-4}$  cm<sup>2</sup>V<sup>-1</sup>s<sup>-1</sup>, an on/off current ratio of  $10^4$ , and a gate current density of  $10^{-6}$  A cm<sup>-2</sup> (ref. 10).

Given the small thickness of the molecular monolayers (about 2.5 nm), a gate current density of  $10^{-6} \text{ A cm}^{-2}$  may seem reasonably low. However, the current density through a molecular monolayer can be less than  $10^{-8} \text{ A cm}^{-2}$  (at 1 V) if no semiconductor is deposited<sup>11</sup>. One possible explanation for the increase in gate current upon deposition of the organic semiconductor is the partial penetration of the SAM by the semiconductor molecules. This penetration is likely to reduce the dielectric thickness and create low-resistance current paths through the SAM. Such interaction between the SAM dielectric and the organic semiconductor is undesirable not only because it leads to increased gate leakage, but also because high-mobility carrier transport in the semiconductor depends critically on a well-defined semiconductor– dielectric interface.

To create dense self-assembled monolayers with sufficient robustness against molecular penetration, we have used a specifically