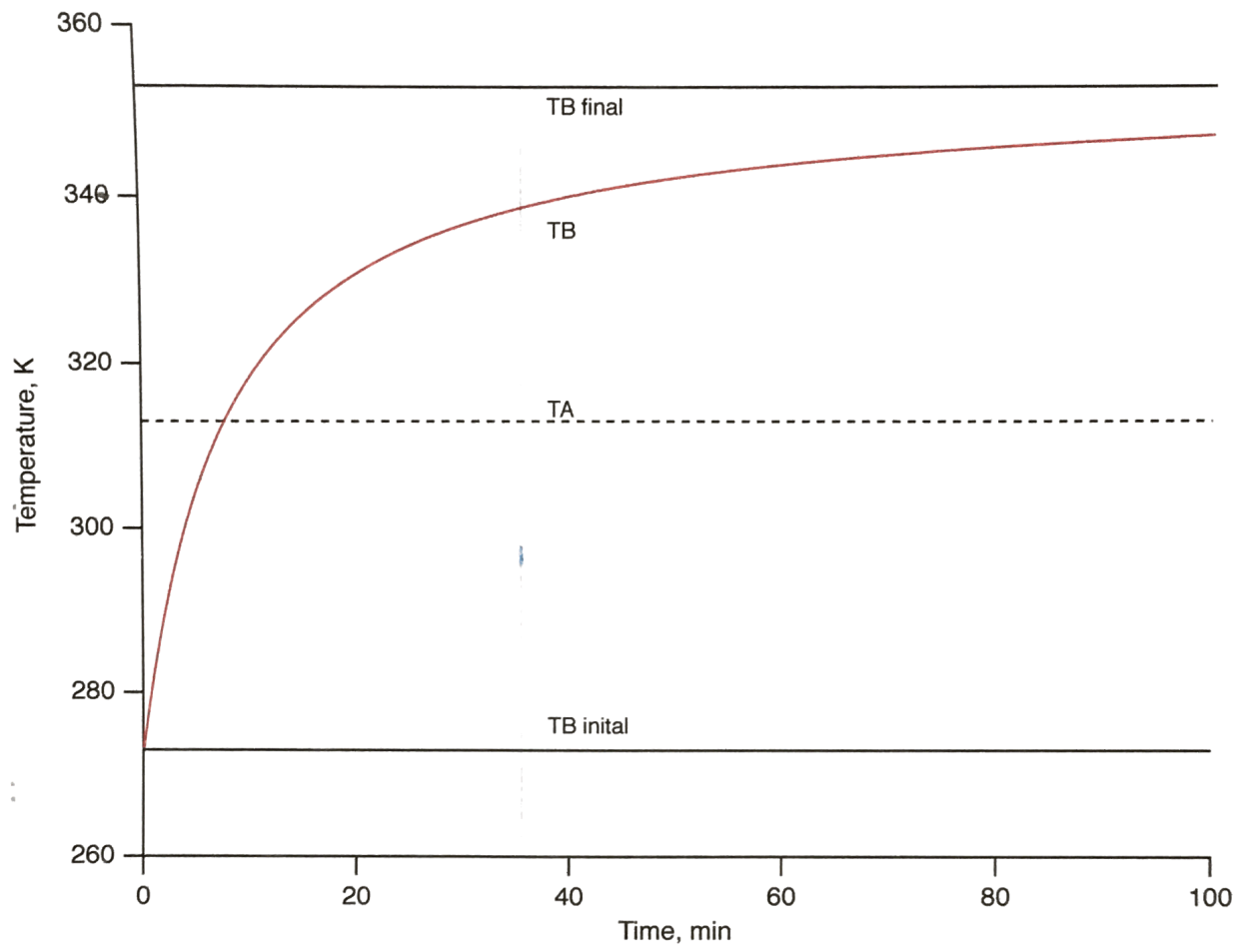


	Answer Sheet	
a)	P_f , MPa	0.794 MPa
	n_f^A , moles	8.18 moles
	n_f^B , moles	6.09 moles
b)	Excess Moles	0.47 moles excess
c)	T_f^B , K	353 K
	P_f^B , MPa	1.38 MPa
	n_f^B , moles	8.19 moles
d)	Function: $T_f^B = A + \frac{B}{Ct + 1}$	
	A, K	353 K
	B, K	80 K
	C, min ⁻¹	0.130 min ⁻¹
e)	Time $T_B = T_A$, min	7.69 min



①

$$(a) \quad n_i = \frac{P_i V}{RT} = \frac{1.38 \text{ MPa} \cdot 2.69 \text{e}^{-4} \text{ cm}^3}{8.31 \frac{\text{MPa cm}^3}{\text{K mole}} \cdot 313 \text{ K}} = 14.3 \text{ moles}$$

$$n_i = n_A^f + n_B^f$$

$$n_A = \frac{P_f \cdot 2.69 \text{e}^{-4} \text{ cm}^3}{8.31 \frac{\text{MPa cm}^3}{\text{K mole}} \cdot T_A}$$

$$n_B = \frac{P_f \cdot 1.74 \text{e}^{-4} \text{ cm}^3}{8.31 \frac{\text{MPa cm}^3}{\text{K mole}} \cdot T_B}$$

$$T_A = 313 \text{ K} \quad T_B = 273 \text{ K}$$

$$n_i = 14.3 \text{ moles} = P_f \left(10.3 \frac{\text{mole}}{\text{MPa}} + 7.67 \frac{\text{mole}}{\text{MPa}} \right)$$

$$P_f = 0.794 \text{ MPa}$$

$$n_A = 0.794 \text{ MPa} \cdot 10.3 \frac{\text{mole}}{\text{MPa}} = 8.18 \text{ moles}$$

$$n_B = 0.794 \text{ MPa} \cdot 7.67 \frac{\text{mole}}{\text{MPa}} = 6.09 \text{ moles}$$

(b)

$$n_A = n_i \frac{V_A}{V_T} = 14.3 \text{ moles} \cdot \frac{2.69 \text{e}^{-4} \text{ cm}^3}{4.43 \text{e}^{-4} \text{ cm}^3} = 8.68 \text{ moles}$$

$$n_B = n_i \frac{V_B}{V_T} = 14.3 \text{ moles} \cdot \frac{1.74 \text{e}^{-4} \text{ cm}^3}{4.43 \text{e}^{-4} \text{ cm}^3} = 5.62 \text{ moles}$$

$$\text{Excess Moles} = 6.09 \text{ moles} - 5.62 \text{ moles} = 0.47 \text{ moles}$$

8.4% excess

For Tank B

$$(c) \quad d(nU) = n_f^B U_f^B = H^A dn = H^A n_f^B = n_f^B (U^A + P U^A) = n_f^B (U^A + R T^A)$$

$$(U_f^B - U^A) = C_v (T_f^B - T^A) = R T^A$$

$$C_p = 8.5 \text{ R}$$

$$C_v = 7.5 \text{ R} \quad \text{i.e.}$$

$$T_f^B = \frac{R T^A}{C_v} + T^A = T^A \left(\frac{R}{C_v} + \frac{C_v}{C_v} \right) = T^A \left(\frac{C_p}{C_v} \right)$$

$$T_f^B = 313 \text{ K} \left(\frac{8.5 \text{ R}}{7.5 \text{ R}} \right) = 353 \text{ K}$$

$$P_f = 1.38 \text{ MPa}$$

$$n_f^B = \frac{P_f V^B}{R T_f^B} = \frac{1.38 \text{ MPa} \cdot 1.74 \text{e}^{-4} \text{ cm}^3}{8.31 \frac{\text{MPa cm}^3}{\text{K mole}} \cdot 353 \text{ K}} = 8.19 \text{ moles}$$

(a)

$$d(nu)^B = H^A dn = n^B du^B + u^B dn$$

$$dn = \left(\frac{dn}{dt}\right) dt \quad \left(\frac{dn}{dt}\right) = 0.1 \frac{\text{mole}}{\text{min}} = \dot{n}$$

$$= \dot{n} dt$$

$$(H^A - u^B) dn = n^B du^B$$

$$(H^A - H^B + (pV)^B) dn = n^B C_v dT^B$$

$$(C_p(T^A - T^B) + RT^B) dn = n^B C_v dT^B$$

$$n^B = n_i^B + \left(\frac{dn}{dt}\right) t$$

both dn^B & n^B are function of time

$$\left(n_i^B + \frac{\dot{n} dt}{\dot{n}}\right) = \frac{C_v dT^B}{C_p T^A + T^B (R - C_p)}$$

$$y = n_i^B + \dot{n} t$$

$$dy = \dot{n} dt$$

$$x = C_p T^A + T^B (R - C_p)$$

$$dx = (R - C_p) dT^B$$

$$\ln \frac{\dot{n} t_f + n_i^B}{\dot{n} t_i + n_i^B} = \frac{C_v}{(R - C_p)} \ln \frac{C_p T^A + T_f^B (R - C_p)}{C_p T^A + T_i^B (R - C_p)}$$

$$t_i = 0$$

$$C_p = C_v + R$$

$$\frac{\dot{n} t_f + n_i^B}{n_i^B} = \left(\frac{C_p T^A + T_f^B (R - C_p)}{C_p T^A + T_i^B (R - C_p)} \right)^{\frac{R - C_p}{C_v}}$$

$$\frac{\dot{n} t_f}{n_i^B} + 1 = \frac{C_p T^A - T_i^B (C_v)}{C_p T^A - T_f^B C_v}$$

$$T_f^B = \frac{C_p T^A}{C_v} + \frac{T_i^B - (C_p/C_v) T^A}{(\dot{n}/n_i^B) t + 1}$$

(3)

$$T_f^B = \left(\frac{8.85R}{7.85R} \right) 313K + \frac{273K - \left(\frac{8.85R}{7.85R} \right) 313K}{\left(\frac{0.1 \text{ mole/min}}{0.767 \text{ mole}} \right) t + 1}$$

$$n_i^B = \frac{PV}{RT} = \frac{0.1 \text{ MPa} \cdot 1.74 \text{ m}^3}{8.31 \frac{\text{J}}{\text{mol K}} \cdot 273K} = 0.767 \text{ mole}$$

$$T_f^B = 353K + \frac{273K - 353K}{\left(0.130 \frac{1}{\text{min}} \right) t + 1}$$

$$T_f^B = 353K - \frac{80K}{\left(0.130 \frac{1}{\text{min}} \right) t + 1}$$

$$T_f^B = 313K \text{ at } 7.69 \text{ min}$$

(using solver in excel)

$$\frac{\left(\frac{80K}{353K - 313K} \right) - 1}{\left(0.130 \frac{1}{\text{min}} \right)} = 7.69 \text{ min}$$