## Quiz 5 Properties of Materials CME 300 October 28, 2011

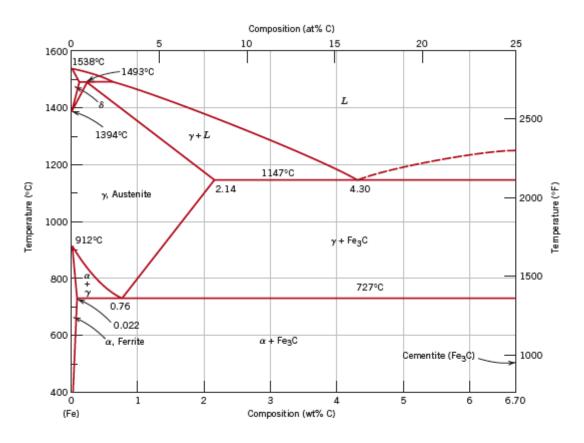
1) Sketch the phase diagram for Iron/Cementite showing the intermetallic phase at 6.70 wt. % carbon, the eutectic at 4.30 wt. % C (1147°C), the peritectic for  $\delta + L \Rightarrow \gamma$  (1493°C), the eutectoid at 0.76 wt. % carbon (727°C), the ferrite miscibility limit at 0.022% and the austenite miscibility limit at 2.14%.

2) Use the phase diagram and sketches where necessary and explain what each of these are:

- a) Pearliteb) Spheroiditec) Martensited) Tempered steele) Bainite
- 3) In a hypoeutectoid steel both eutectoid and proeutectoid ferrite exits. Explain the difference between them. What is the concentration of carbon in each (use your phase diagram)?
- 4) Explain the composition and heat treatment, microstructure and mechanical characteristics of gray cast iron.

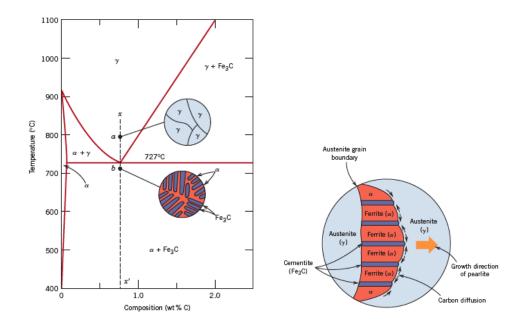
## ANSWERS: Quiz 5 Properties of Materials CME 300 October 28, 2011

1) Sketch the phase diagram for Iron/Cementite showing the intermetallic phase at 6.70 wt. % carbon, the eutectic at 4.30 wt. % C (1147°C), the peritectic for  $\delta + L \Rightarrow \gamma$  (1493°C), the eutectoid at 0.76 wt. % carbon (727°C), the ferrite miscibility limit at 0.022% and the austenite miscibility limit at 2.14%.



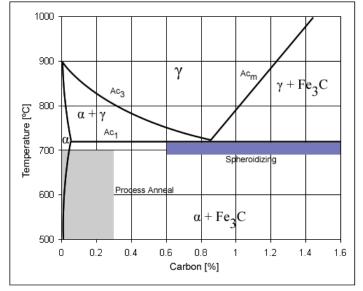
2) Use the phase diagram and sketches where necessary and explain what each of these are:a) Pearlite

Pearlite is a eutectoid morphology formed as shown in the following phase diagram.





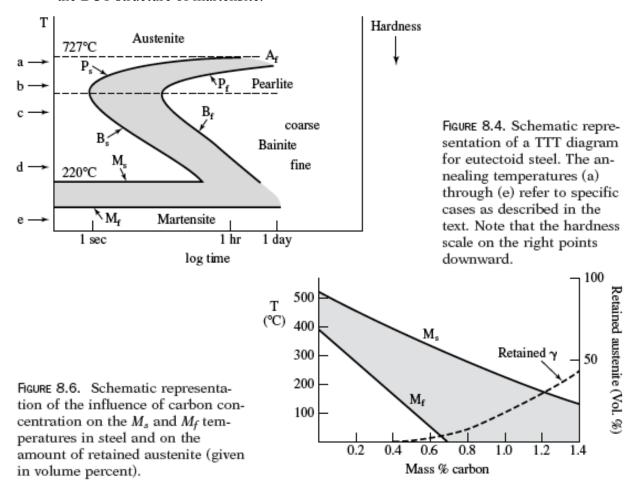
Spheroidite forms when carbon steel is heated to approximately 700 °C for over 30 hours. Spheroidite can form at lower temperatures but the time needed drastically increases, as this is a diffusion-controlled process. The result is a structure of rods or spheres of cementite within primary structure (ferrite or pearlite, depending on which side of the eutectoid you are on). The purpose is to soften higher carbon steels and allow more formability. This is the softest and most ductile form of steel.



c) Martensite

Martensite is achieved by a rapid quench from austenite steel to temperatures below

500°C in a diffusionless process that converts the FCC structure of austenite to the BCT structure of martensite.

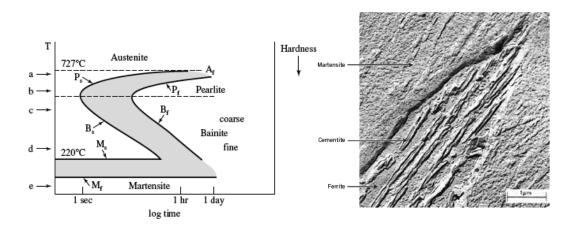


## d) Tempered steel

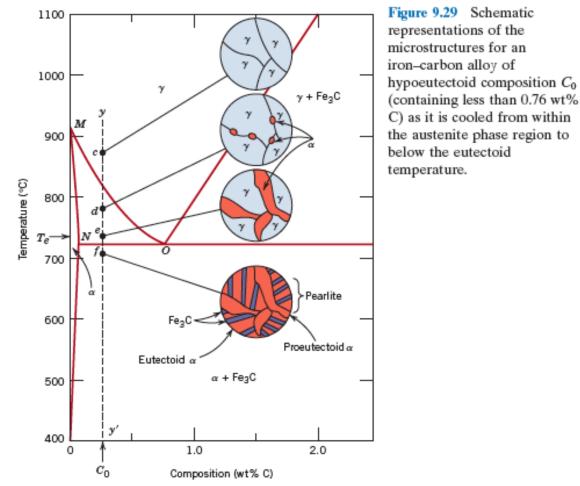
Since martensite is a brittle material it is desirable to controllably revert the martensitic structure into ferrite and cementite phases by annealing the steel just below the eutectoid temperture. This process is called tempering and the resulting steel is tempered steel.

e) Bainite

Moderate quenches lead to a needle like cementite in a ferrite matrix dispersed in austenite or martensite structure called bainite.



3) In a hypoeutectoid steel both eutectoid and proeutectoid ferrite exits. Explain the difference between them. What is the concentration of carbon in each (use your phase diagram)?
Proeutectoid ferrite forms above the eutectoid temperature and has a final equilibrium composition of 0.022 % C. Eutectoid ferrite has a slightly lower concentration of carbon since it forms below the eutectoid temperature and exists in the pearlite structure.



4) Explain the composition and heat treatment, microstructure and mechanical characteristics of gray cast iron.

Grey cast iron is 2.5 to 4 wt. % carbon and 1 to 3 % silicon. Microstructure includes flakes of graphite surrounded by ferrite or pearlite matrix. Grey cast iron is weak and brittle under tension but is strong under compression. Good damping for vibrations, easily cast, low cost.

