Aluminium and its alloys

Alumina raw materials

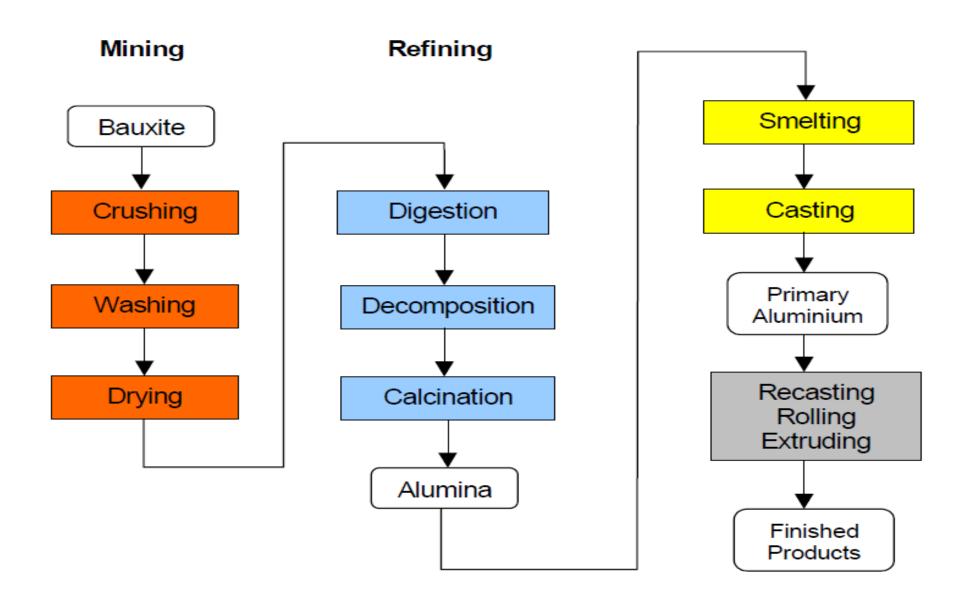
Alumina can be processed from bauxite, kaolinite and nepheline



Nepheline

30% Alumina (Al₂O₃) 40% Silica (SiO₂) 20% Na₂O + K₂O

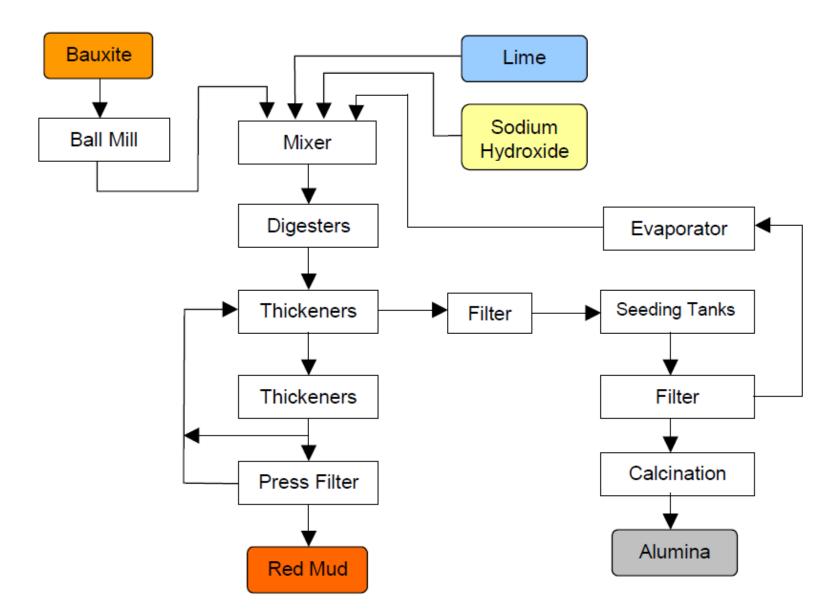
Bayer Process



Bayer Process

- Bauxite is washed ground and dissolved in caustic soda (NaOH) at high pressure and temperature
- Sodium aluminate solution containing nonsoluble bauxite residues sink to the bottom red mud.
- Seeding agent is added to the clear sodium aluminate solution to give alumina precipitates
- Precipitates are passed through a rotary kiln for calcination at ~1100 oC to give white powder of pure alumina.

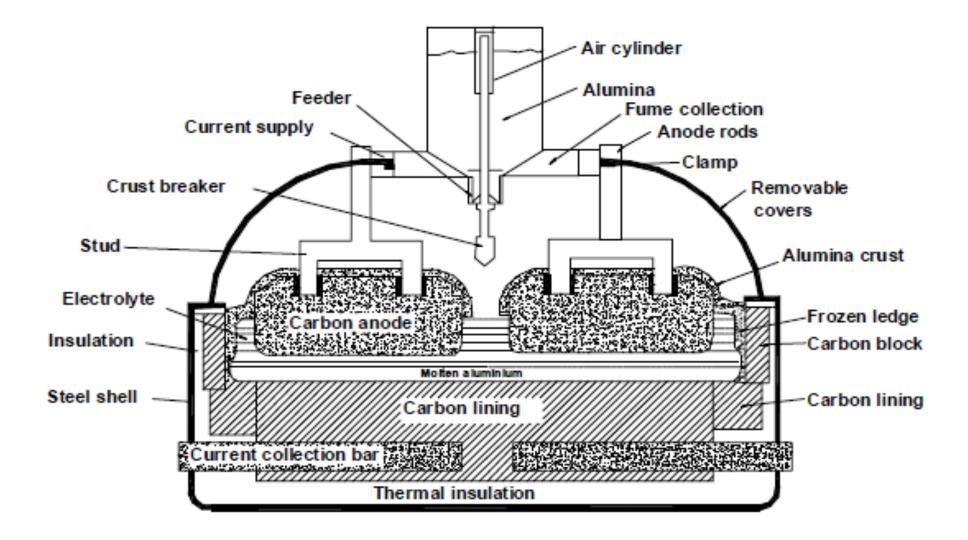
Bayer Process



(Aluminium smelting/electrolysis) Feed Material: Alumina (Al₂O₃) Electrolyte: Cryolite (Na₃AlF₆) Anode material: Carbon Cathode material: Carbon/graphite

Anode: $2O^{2-} \rightarrow O_2 + 4e$ - $2O^{-2} + C \rightarrow CO_2 + 4e$ -Cathode: $Al_3 + 3e - \rightarrow Al$ Overall Rxn: $2Al_2O_3 + 3C \rightarrow 4Al + 3CO_2$

Hall-Héroult process



Properties of Aluminium

- High corrosion resistance
- Excellent machining properties
- Light weight
- High thermal/electrical conductivity
- High ductility/easily deformable

Wrought Aluminium alloys

Composition of aluminium alloys are regulated by internationally agreed classifications system

- 1XXX AI of 99% minimum purity
- 2XXX AI Cu alloys
- 3XXX AI Mn alloys
- 4XXX AI Si alloys
- 5XXX AI Mg alloys
- 6XXX AI Mg Si alloys
- 7XXX AI Zn Mg alloys
- 8XXX Miscellaneous alloys, e.g. aluminiumlithium alloys

Main groups of wrought aluminium

- 1xxx series (Super-purity and commercial-purity aluminium)
- 3xxx series (Al-Mn and Al-Mn-Mg alloys)
- 5xxx series (Al-Mg alloys)
- 8xxx series (Miscellaneous alloys

- 2xxx series (Al-Cu and Al-Cu-Mg alloys)
- 6xxx series (Al-Mg-Si alloys)
- 7xxx series (Al-Zn-Mg and Al-Zn-Mg-Cu alloys

Main groups of wrought aluminium

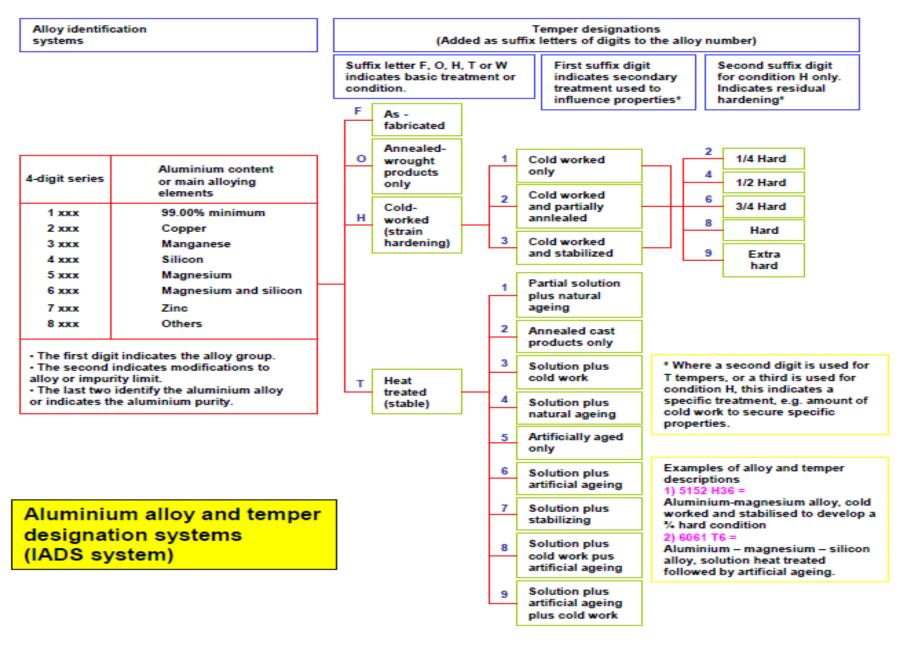
Non-heat-treatable alloys

- 1xxx series (Super-purity and commercial-purity aluminium)
- 3xxx series (AI-Mn and AI-Mn-Mg alloys)
- 5xxx series (AI-Mg alloys)
- 8xxx series (Miscellaneous alloys

Heat-treatable alloys

- 2xxx series (AI-Cu and AI-Cu-Mg alloys)
- 6xxx series (AI-Mg-Si alloys)
- 7xxx series (AI-Zn-Mg and AI-Zn-Mg-Cu alloys

Aluminium alloys and temper designations



1xxx series Properties:

- Low tensile strength (90 MPa)
- Yield stress of 7-11 MPa.

- Electrical conductors
- Chemical process equipment
- Foils
- Decorative finishes
- Capacitor (by panasonic)





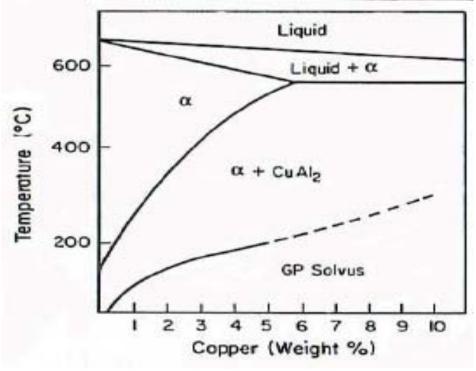
2xxx series

Properties:

- High strength (2119: σTS 505 MPa).
- Good creep strength at high temp.
- High toughness at cryogenic temp.
- Good machinability.

- Welding wires
- Fuel Tanks
- Aircraft body





3xxx series

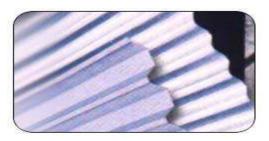
Al-Mn alloys (upto 1.25% Mn) Greater amount leads to large primary Al6Mn particles) deleterious local ductility)

Properties:

- Moderate strength, i.e., σTS ~
 110 MPa in annealed 3003
- High ductility
- Excellent corrosion resistance

Applications:

- Foil
- Roofing sheet



Al-Mn-Mg alloys

(provide solid solution strengthening) and widely used in a variety of strain hardened tempers

Properties:

- Moderate strength, i.e., σTS ~ 180 MPa in annealed 3004.
- Readily fabricated
- Excellent corrosion resistance

Applications:

Manufacturing beverage cans



4xxx series

Aluminum / Silicon alloys (Silicon ranging from 0.6% to 21.5%)

Properties:

- Excellent weldability and fair weld strength of 120 MPa
- Moderate strength
- Has heat and non-heat-treatable properties
- Excellent corrosion resistance

Applications:

- Used as filler material
- Welding and brazing wire
- Forged engine pistons

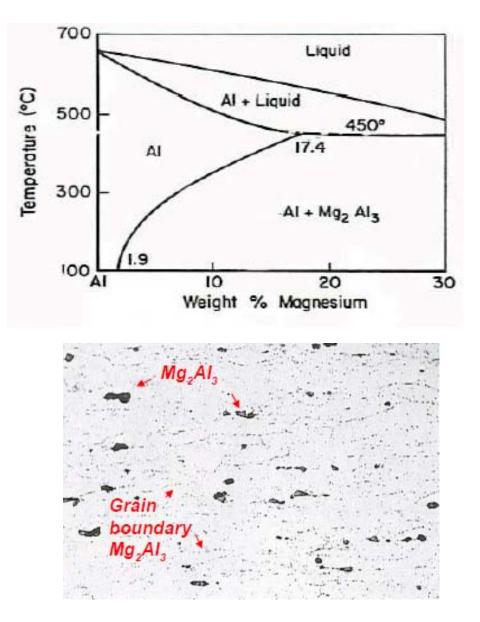
Main application: Architectural applications

5xxx series

Properties:

- AI-0.8Mg (5005):σy 40 MPa, σTS 125 MPa
- Al-(4.7-5.5)Mg (5456): σy 160, σTS 310 MPa
- High rate of work hardening
- High corrosion resistance
- Bright surface finish

- Transportation structural plates
- Large tanks for petrol, milk, grain
- Pressure vessel
- Architectural components



6xxx series

Properties:

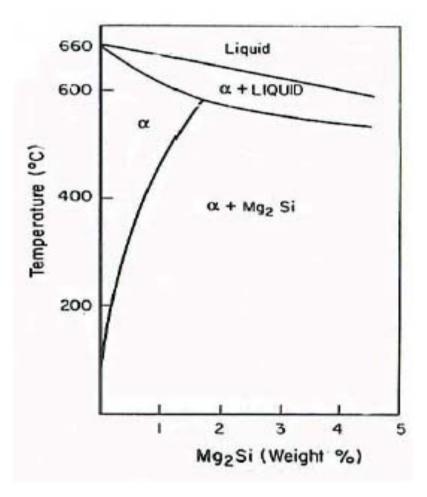
- Medium-strength structural alloys (most widely used 6063-T6, σy 215 MPa, σTS 245)
- Higher strength on ageing, 6013 -Al-Mg-Si-Cu, σy 330 MPa(T6) and 415 (MPa) T8.

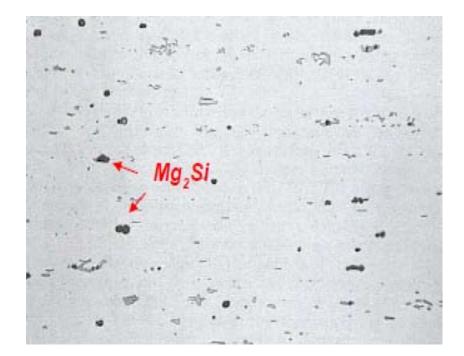
- Transportation structural plates
- Large tanks for petrol, milk, grain
- Pressure vessel
- Architectural components





6xxx series





7xxx series

Properties:

- Strength is insensitive to cooling rate hence suitable for welding
- Yield strength might be double to Al-Mg and Al-Mg-Si alloys (~ upto 600 MPa)
- Stress corrosion cracking resistance in Al-Zn-Mg-Cu alloys



AI 7039 aircraft construction

- Light weight military bridge
- Aircraft construction



Al 7075 Component in motorcycle Al 7005 post box

8xxx series

Properties:

- High corrosion resistance at
- high temp & pressure
- Deep drawing

- AI-1.1Ni-0.6Fr (8001) nuclear energy installations
- Al-0.75Fe-0.7Si (8011) bottle caps.
- AI-Sn (up to 7%) soft bearings
- AI-Li for aerospace applications

Designations of cast aluminium alloys

United States Aluminium Association system (Using four-digit system)

- 1xx.x AI, 99.00% or greater AI alloys grouped by major alloying elements
- 2xx.x Cu
- 3xx.x Si with added Cu and/or Mg
- 4xx.x Si
- 5xx.x Mg
- 7xx.x Zn
- 8xx.x Sn
- 9xx.x Other elements
- 6xx.x Unused series

1xx.x series

- Second two digits indicate the minimum percentage of AI, Eg: 150.x = 99.50% AI.
- Last digit (after decimal point) indicates product forms.
- 1 = casting, 2 = ingot

2xx.x to 9xx.x series

- Second two digits identify the different aluminium alloys
- Last digit (after decimal

point) indicates product forms

Cast aluminium alloys

Properties required for good casting

- Low melting temperature
- Low solubility of gases except H2
- Good fluidity
- Good surface finishes

Main disadvantage

High solidification shrinkage (3.5-8.5%)

Factors controlling properties

- Melting and pouring practices
- Impurity levels
- Grain size
- Solidification rate



Cast aluminium alloys are widely used for transport applications, Eg: Cast engine block

Strengthening Mechanism of Metals (Solid Solution Strengthening)

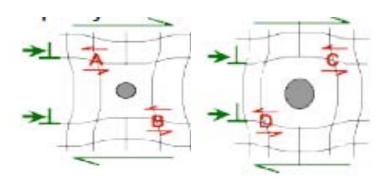
Adding other elements in solid solution

Mechanism:

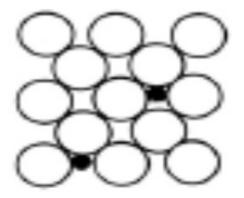
Dissolved impurities *distort lattice* by

Substitutional / Interstitial

- strengthening effect increases as
- $|\Delta r|\uparrow (\Delta r = rhost rimpurity)$
- The stress generated can produce a barrier to dislocation motion



Smaller and bigger substitional impurity (atom)



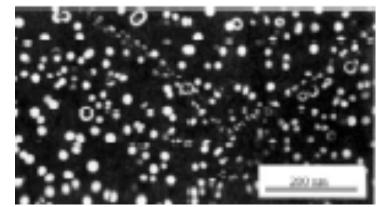
Impurities (atoms) occupying interstitial positions

Strengthening Mechanism of Metals (Precipitation(Age Hardening)/dispersion hardening)

Adding second phase particles or precipitation of supersaturated solid solution

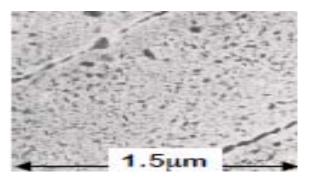
Mechanism:

 dislocation movement is impeded across grain boundaries between different phases

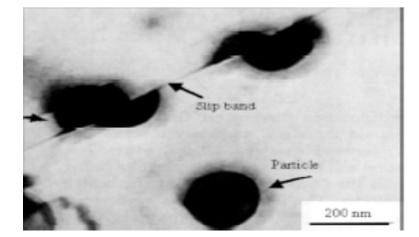


Al₃Li precipitates

Example is Al-4%Cu alloy



Al₂Cu₃ precipitates at grain boundaries



Strengthening Mechanism of Metals (Strengthening by Grain Size Reduction)

Strengthening by reduction in grain size

Mechanism:

- In general, slip across grain boundary involves
- Discontinuity of slip planes
- Change in slip direction
- For many materials, the yield strength increases with a decrease in grain size

 The yield strength and the grain size are related by the

Hall-Petch Equation

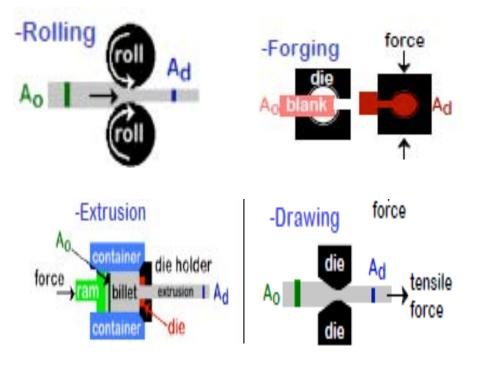
 $\sigma_{yield} = \sigma_0 + k_y d^{-1/2}$

Strengthening Mechanism of Metals (Strain Hardening)

Cold work (strengthening by lowtemperature plastic deformation)

Mechanism:

- Plastic deformation creates dislocations
- Upon repeated or extensive deformation, dislocations multiply, move, and (on average) repel each other thereby decreasing dislocation mobility
- This increases the yield strength and the ultimate tensile strenght



References