## Quiz 9 Polymer Properties October 24, 2014

a) The scattering Function for a sphere is given by:

$$I(q) = 9G \left[ \frac{\sin qR - qR\cos qR}{\left(qR\right)^3} \right]^2 \tag{1}$$

(for example see Pedersen, J. S. (1997). *Adv. Colloid Interface Sci.* **70**, 171-210 or RJ Roe's text).

where R is the radius of the sphere, G is proportional to the square of the volume of the sphere times the density squared. Show that equation (1) agrees with Guinier's Law at low-q, and with Porod's Law at high-q ( $I(q) \sim Sq^{-4}$  where S is the sphere surface area).

$$\sin x = x - \frac{1}{6}x^3 + \frac{1}{120}x^5 - \frac{1}{5040}x^7 + \dots \text{ for } -\infty < x < \infty$$

$$\cos x = 1 - \frac{1}{2}x^2 + \frac{1}{24}x^4 - \frac{1}{720}x^6 - \dots \text{ for } -\infty < x < \infty$$

 $\langle \cos^2 \theta \rangle = \langle \sin^2 \theta \rangle = \frac{1}{2}$ , and  $\langle \sin \theta \cos \theta \rangle = 0$ 

- b) Show that the radius of gyration for equation (1) agrees with the radius of gyration for a sphere. (Calculate the radius of gyration for a sphere in terms of R then compare with what you obtain by equating the low-q extrapolation of (1) with Guinier's Law and solving for  $R_{\rm g}$ .)
- c) The Debye-Bueche function is often used to describe scattering from solid objects of unknown structure.

$$p(r) = K \exp\left(-\frac{r}{\xi}\right)$$
 Debye-Bueche Function  $I(q) = \frac{G}{1 + q^4 \xi^4}$  (2)

Critique equation (2) by comparison with Gunier's Law.

Does equation (2) follow Porod's law at high-q? (Is the prefactor proportional to S?)

d) Debye derived equation (3) for polymers in dilute solutions,

$$g(q)_{Gaussian} = \frac{2}{Q^2} [Q - 1 + \exp(-Q)]$$

where 
$$Q = q^2 N b^2 / 6 = q^2 R_g^2$$
 (3)

Can this equation be used for polymers in dilute solution? Explain why by extrapolation to high-q.

Can equation (3) be used for polymer chains in a melt?

e) The radius of gyration for a polymer is equal to the end-to-end distance divided by  $\sqrt{6}$  for a Gaussian chain. How is the hydrodynamic radius related to the end-to-end distance? (You may need to give a structural picture of the hydrodynamic radius to answer this.)

## ANSWERS: Quiz 9 Polymer Properties October 24, 2014

1) a) Substituting the power series for sin and cos we obtain:

$$\begin{array}{c}
y = gh \\
q G \left[ \frac{x - \frac{x^{3}}{3!} + \frac{x^{5}}{5!} - \left[ x - \frac{x^{3}}{2!} + \frac{x^{5}}{4!} - \frac{x^{7}}{6!} \right] \right]^{2} \\
= q G \left[ \frac{1}{x^{3}} + \frac{1}{5!} - \left[ x - \frac{x^{3}}{2!} + \frac{x^{5}}{4!} - \frac{x^{7}}{6!} \right] \right]^{2} \\
= G \left[ 1 - \frac{x^{2}}{30} \right]^{2} = G \left[ erp \left( \frac{e^{2}h_{0}^{2}}{5!} \right) \right]^{2} \\
= G \left[ erp - \frac{e^{2}h_{0}^{2}}{5!} \right]^{2} \\
= G \left[ erp - \frac{e^{2}h_{0}^{2}}{5!$$

At high-q you expand the squared term and find the average values for the trig terms at high-q:  $\langle\cos^2\theta\rangle = \frac{1}{2}$ ;  $\langle\sin^2\theta\rangle = \frac{1}{2}$ ;  $\langle\cos\theta\sin\theta\rangle = 0$ . So the function yields  $9G/(2R^4q^4)$ . G is proportional to  $V^2$  or  $R^6$ , so we have  $S \sim R^2$ .

- b) For a sphere we take the integral of R4 divided by the integral of R2 to obtain  $R_g^2 = 5/3$  R<sup>2</sup>. The low-q extrapolation of equation (1) yields the same answer.
- c) Equation (2) is the first two terms of G exp(-q<sup>4</sup> $\xi^4$ ) which is not the same form as Guinier's Law so the function can not be correct. At high-q, using G ~ V<sup>2</sup>, yields R<sup>6</sup>/ $\xi^4$  which has the units of area, but the correlation length is not directly related to the surface to volume ratio so the expression is muttled.
- d) Equation (3) can only be used for Gaussian polymers so it can not be used for polymers in dilute solution which display good solvent scaling. I can be used for polymers in the melt.
- e) There is not good direct relationship between the hydrodynamic radius and the end-toend distance of a polymer chain. It depends on the degree of drainage of the coil for one thing.